

AP Chemistry Final Study Guide

Unit 1: Atomic Structure and Properties (7%–9%)

- At STP, one mole of gas always occupies 22.4 L
- **Mass Spectroscopy:**
 - Atoms have different isotopes (carbon-12, carbon-14, etc) and this will affect the average atomic mass
 - The most abundant isotope will have the greatest effect
- **Coulomb's Law:**
 - The greater the charge of the nucleus, the more energy is required to ionize an electron
 - The closer an electron is to the nucleus, the more energy is required to ionize an electron
- **Electromagnetic Radiation:**
 - When atoms absorb energy, electrons jump to higher energy levels
 - When electrons drop from higher to lower energy levels, atoms give off electromagnetic radiation energy
- **Kinetic Energy of Ionized Electrons:**
 - The faster an ejected electron is going, the more kinetic energy it has
 - Electrons originally further away from the nucleus require less energy to eject, so it will have more kinetic energy
- **Electron Configuration:**
 - When entering the d-block (transition metals), the energy level drops by 1 unless you are using shorthand notation
 - Nickel: $1s^2 2s^2 2p^6 3s^2 3p^6 3d^8 4s^2$ (notice the d-block)
 - Nickel (with shorthand): $[\text{Ar}] 4s^2 3d^8$
- **Law of Definite Proportions:**
 - A pure chemical compound broken down into elements always contains elements of a fixed ratio, independent of where and how it was created
- **Law of Multiple Proportions:**
 - When elements form compounds, the proportions of the elements in those chemical compounds can be expressed in small whole number ratios
- **Aufbau Principle:**
 - When building up the electron configuration of an atom, electrons are placed in orbitals, subshells, and shells in order of increasing energy

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Ap Chemistry Study Guide:

AP Chemistry Study Guide, 2018 [CliffsNotes AP Chemistry](#) Bobrow Test Preparation Services, 2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter's end Two full length practice tests with detailed answer explanations are included in the book **AP Chemistry Exam Study Guide** Accepted, 2019-09-19 *AP Chemistry 2020 & 2021: AP Chemistry Review Book and Practice Questions for the Advanced Placement Chem Exam* Test Prep Books, 2019-11-21 Test Prep Books AP Chemistry 2020 2021 AP Chemistry Review Book and Practice Questions for the Advanced Placement Chem Exam Made by Test Prep Books experts for test takers trying to achieve a great score on the AP Chemistry exam This comprehensive study guide includes Quick Overview Find out what's inside this guide Test Taking Strategies Learn the best tips to help overcome your exam Introduction Get a thorough breakdown of what the test is and what's on it Atomic Structure and Properties Molecular and Ionic Compound Structure and Properties Intermolecular Forces and Properties Chemical Reactions Kinetics Thermodynamics Equilibrium Acids and Bases Applications of Thermodynamics Practice Questions Practice makes perfect Detailed Answer Explanations Figure out where you went wrong and how to improve AP R and Advanced Placement R are trademarks registered by the College Board which is not affiliated with and does not endorse this product Studying can be hard We get it That's why we created this guide with these great features and benefits Comprehensive Review Each section of the test has a comprehensive review created by Test Prep Books that goes into detail to cover all of the content likely to appear on the ISEE Lower Level test Practice Test Questions We want to give you the best practice you can find That's why the Test Prep Books practice questions are as close as you can get to the actual AP Chemistry exam Answer Explanations Every single problem is followed by an answer explanation We know it's frustrating to miss a question and not understand why The answer explanations will help you learn from your mistakes That way you can avoid missing it again in the future Test Taking Strategies A test taker has to understand the material that is being covered and be familiar with the latest test taking strategies These strategies are necessary to properly use the time provided They also help test takers complete the test without making any errors Test Prep Books has provided the top test taking tips Customer Service We love taking care of our test takers We make sure that you interact with a real human being when you email your comments or concerns Anyone planning to take this exam should take advantage of this Test Prep Books study guide Purchase it today to receive access to AP Chemistry review materials AP Chemistry practice questions Test taking strategies [CliffsAP 5 Chemistry Practice Exams](#) Gary S Thorpe, 2007-05-03 Your complete guide to a higher score on the AP Chemistry exam Why CliffsAP Guides Go with the name you know and trust Get the information you need fast Written by test prep specialists About the contents Introduction Describes the exam's format Discusses the topics covered Gives proven strategies for answering the multiple choice and free response questions Answers FAQs about the exam 5 Full length AP

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