

## Electron Configurations Worksheet - Answers

Write the complete ground state electron configurations for the following:

1)	lithium	$1s^2 2s^1$
2)	oxygen	$1s^2 2s^2 2p^4$
3)	calcium	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2$
4)	titanium	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^2$
5)	rubidium	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^1$
6)	lead	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2 4f^{14} 5d^{10} 6p^2$
7)	erbium	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2 4f^{12}$

Write the abbreviated ground state electron configurations for the following:

8)	helium	$1s^2$ (this one cannot be abbreviated)
9)	nitrogen	$[\text{He}] 2s^2 2p^3$
10)	chlorine	$[\text{Ne}] 3s^2 3p^5$
11)	iron	$[\text{Ar}] 4s^2 3d^6$
12)	zinc	$[\text{Ar}] 4s^2 3d^{10}$
13)	barium	$[\text{Xe}] 6s^2$
14)	polonium	$[\text{Xe}] 6s^2 4f^{14} 5d^{10} 6p^4$

## 5 3 Practice Electron Configuration Worksheet Answers

**Scott C. Dulebohn**

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