

**CHAPTER 4 REVIEW*****Arrangement of Electrons in Atoms*****SECTION 1****SHORT ANSWER** Answer the following questions in the space provided.

1. In what way does the photoelectric effect support the particle theory of light?

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2. What is the difference between the ground state and the excited state of an atom?

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3. Under what circumstances can an atom emit a photon?

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4. How can the energy levels of the atom be determined by measuring the light emitted from an atom?

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5. Why does electromagnetic radiation in the ultraviolet region represent a larger energy transition than does radiation in the infrared region?

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## 5 2 Section Review Electrons Arrangement In Atoms

**Karin Nielsen-Saines**



## 5 2 Section Review Electrons Arrangement In Atoms:

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