

(d) The halogen fluorine is too reactive to use in a school or college laboratory. Fluorine is a powerful oxidising agent. It will react with water as shown below.

(i) Complete and balance the equation for the reaction of fluorine with water.



(ii) Using oxidation numbers, show what has been oxidised and what has been reduced in this reaction.

.....  
 .....  
 .....  
 ..... [2]

(e) Fluorine will react violently with gallium to produce gallium fluoride.

Mendeleev originally called gallium 'eka-aluminium' as he predicted that gallium would have similar properties to aluminium.

(i) Complete the electron structure of the gallium atom.

1s<sup>2</sup> ..... [1]

(ii) Use Mendeleev's prediction to suggest the empirical formula of gallium fluoride.

..... [1]

[Total: 19]

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