

Section A [15 marks]

Answer all questions in this section.

1. Tetraethyllead(IV), $(\text{CH}_3\text{CH}_2)_4\text{Pb}$, is added to petroleum as an anti-knock agent. $(\text{CH}_3\text{CH}_2)_4\text{Pb}$ is prepared from ethane, CH_3CH_3 , according to the following reactions:



Which statement about the reactions is not true?

A. One mole of Cl_2 produces one mole of $\text{CH}_3\text{CH}_2\text{Cl}$.
 B. Four moles of Cl_2 produce one mole of $(\text{CH}_3\text{CH}_2)_4\text{Pb}$.
 C. One mole of HCl is formed for each mole of $(\text{CH}_3\text{CH}_2)_4\text{Pb}$ produced.
 D. Four moles of CH_3CH_3 is required for each mole of $(\text{CH}_3\text{CH}_2)_4\text{Pb}$ produced.

2. Gas X is collected in a 250 cm^3 volumetric flask at 25°C until the pressure of the gas is 73.5 kPa . The mass of X in the flask is 0.118 g at 25°C . What is the mass of 1 mol of X ?
 [The gas constant, R , is $8.31 \text{ J K}^{-1} \text{ mol}^{-1}$.]

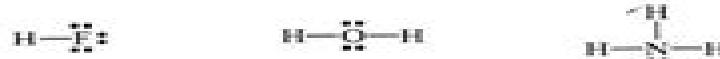
A. 7.42 g B. 11.3 g C. 14.6 g D. 15.9 g

3. An atom M has seven valence electrons and forms a stable M^{2+} ion in an aqueous solution. What is the electronic configuration of atom M ?

A. $1s^2 2s^2 2p^6 3s^2 3p^2$
 B. $1s^2 2s^2 2p^6 3s^2 3p^6 3d^0$
 C. $1s^2 2s^2 2p^6 3s^2 3p^6 3d^0 4s^2$
 D. $1s^2 2s^2 2p^6 3s^2 3p^6 3d^7 4s^2$

4. Which compound has an ionic bond?
 A. HCl B. BeCl_2 C. AlBr_3 D. $\text{C}_2\text{H}_5\text{NH}_2\text{Br}$

5. The Lewis structures of molecules HF , H_2O and NH_3 are shown respectively below.



Which statement about the molecules is true?

A. HF and H_2O molecules are planar.
 B. H_2O molecule is a stronger Lewis base than NH_3 molecule.
 C. The boiling points of the molecules increase in the order of H_2O , HF , NH_3 .
 D. The hydrogen bonds formed by the molecules decrease in the order of HF , H_2O , NH_3 .

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