

## IB ESS EXAM NOTES

### TOPIC 1: Systems and Models

- **Systems:** an assemblage of parts and their relationship forming a functioning entirety or whole
  - o Open systems: exchanges matter and energy
  - o Closed systems: exchanges only energy
  - o Isolated systems: neither matter nor energy and is theoretical

#### - **Laws of thermodynamics**

- o 1st: energy is neither created nor destroyed, only changes forms
- o 2nd: the entropy of a closed system increases; when energy is transformed into work, some energy is always lost as waste heat

### **Equilibrium**

- o Steady-state: in open systems, continuous inputs and outputs of energy and matter, system as a whole remains in a constant state, no long term changes.
- o Static: no change over time; when the state of equilibrium is distributed, the system adapts a new equilibrium; can't occur in living systems
- o Stable: the system returns to the same equilibrium after disturbances
- o Unstable: system returns to a new equilibrium after disturbances

#### - **Feedback**

- o Positive: results in a further decrease of output and the system is destabilized and pushed into a new state of equilibrium
- o Negative: tends to neutralize or counteract any deviation from an equilibrium and tends to stabilize systems

#### - **Transfers and transformations**

- o Transfers:
  - The movement of material through living organisms
  - Movement of material in non-living process
  - The movement of energy

#### o **Transformations**

- Matter to matter
- Energy to energy
- Matter to energy
- Energy to matter

#### - **The Gaia model**

# 13 Ib Ess Exam Paper Nov

**Émile Littré**



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