

Abstract

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Clemens Wendtner



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$$2\text{ NaCl} + \text{F}_2 \rightarrow 2\text{ NaF} + \text{Cl}_2$$

$$2\text{ H}_2\text{O} + 2\text{ H}_2\text{O}_2 \rightarrow 2\text{ H}_2\text{O} + \text{O}_2$$

$$\text{Pb(OH)}_2 + 2\text{ HCl} \rightarrow 2\text{ H}_2\text{O} + \text{PbCl}_2$$

$$2\text{ AlBr}_3 + 3\text{ K}_2\text{SO}_4 \rightarrow 6\text{ KBr} + \text{Al}_2\text{SO}_4$$

$$\text{CH}_4 + 2\text{ O}_2 \rightarrow \text{CO}_2 + 2\text{ H}_2\text{O}$$

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$$\text{N}_2 + \text{F}_2 \rightarrow \text{NF}_3$$

$$2\text{ C}_6\text{H}_{10} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$$

$$3\text{ HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$$

$$\text{GaBr}_3 + \text{NaBr} \rightarrow \text{SNO}_4 + \text{NaSO}_3 + \text{Ga}_2\text{SO}_3 + 3\text{ NF}_3 + \text{SnF}_2 + \text{N}_2\text{O}_3$$
 using the equation from problem 2 above answer the following

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$$\text{N}_2 + 3\text{ F}_2 \rightarrow 2\text{ NF}_3$$

$$2\text{ C}_6\text{H}_{10} + 17\text{ O}_2 \rightarrow 12\text{ CO}_2 + 10\text{ H}_2\text{O}$$

$$3\text{ HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$$

$$2\text{ GaBr}_3 + 3\text{ Na}_2\text{SO}_3 \rightarrow \text{Ga}_2\text{SO}_3 + 6\text{ NaBr} + 3\text{ SNO}_2 + 3\text{ NF}_3 + \text{SnF}_2 + \text{N}_2\text{O}_3$$
 using the

following equation $2 \text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2 \text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

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 1 $\text{N}_2 + 3 \text{F}_2 \rightarrow 2 \text{NF}_3$
 2 $2 \text{C}_6\text{H}_{10} + 17 \text{O}_2 \rightarrow 12 \text{CO}_2 + 10 \text{H}_2\text{O}$
 3 $\text{HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$
 4 $2 \text{GaBr}_3 + 3 \text{Na}_2\text{SO}_3 \rightarrow \text{Ga}_2\text{SO}_3 + 3 \text{NaBr}$
 5 $3 \text{SnO}_2 + \text{N}_2\text{F}_6 \rightarrow \text{SnF}_2 + \text{N}_2\text{O}_3$
 using the following equation
 2 $\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2 \text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

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 1 $2 \text{N}_2 + 3 \text{F}_2 \rightarrow 2 \text{NF}_3$
 2 $\text{CaH}_2 + 2 \text{H}_2\text{O} \rightarrow \text{Ca(OH)}_2 + \text{H}_2$
 3 $\text{HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$
 4 $2 \text{GaBr}_3 + 3 \text{Na}_2\text{SO}_3 \rightarrow \text{Ga}_2\text{SO}_3 + 3 \text{NaBr}$
 5 $3 \text{SnO}_2 + \text{N}_2\text{F}_6 \rightarrow \text{SnF}_2 + \text{N}_2\text{O}_3$

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 $\text{SiO}_2 + 3 \text{C} \rightarrow \text{SiC} + 2 \text{CO}$

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 1 $\text{N}_2 + 3 \text{F}_2 \rightarrow 2 \text{NF}_3$
 2 $2 \text{C}_6\text{H}_{10} + 17 \text{O}_2 \rightarrow 12 \text{CO}_2 + 10 \text{H}_2\text{O}$
 3 $\text{HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$
 4 $2 \text{GaBr}_3 + 3 \text{Na}_2\text{SO}_3 \rightarrow \text{Ga}_2\text{SO}_3 + 3 \text{NaBr}$
 5 $3 \text{SnO}_2 + \text{N}_2\text{F}_6 \rightarrow \text{SnF}_2 + \text{N}_2\text{O}_3$
 solve the following stoichiometry grams problems 6

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