

Section I Part A – Multiple Choice No Calculator
28 Questions – 50 Minutes

Directions: Read each question carefully and select only one choice that best answers each question. Mark your answers on the answer sheet, if provided. All other work besides your answer is not graded. Your total score is based on the number of questions correct, with no points being deducted for incorrect or unanswered questions. A calculator is not allowed on this section of the exam.

1. In the odd function $f(x)$, the $\lim_{x \rightarrow -2} f(x) = -\infty$. Which one of the following must also be true?
- a. $\lim_{x \rightarrow -2} f(x) = -\infty$ c. $\lim_{x \rightarrow -2} f(x) = -\infty$
 b. $\lim_{x \rightarrow -2} f(x) = \infty$ d. $\lim_{x \rightarrow -2} f(x) = \infty$
2. If the functions $f(x) = \sqrt{x}$ and $g(x) = \frac{1}{x-2}$ were combined according to the descriptions below, then which of the following would the combined function have a domain of $[0, 2) \cup (2, \infty)$?
- i. $f(x) + g(x)$
 ii. $(g \circ f)(x)$
 iii. $\frac{f(x)}{g(x)}$
- a. i only c. i and iii only
 b. ii only d. i, ii, and iii

For questions 3–4, use the graph of $g(x)$ on the right.

3. Which one of these intervals could g have an average rate of change of -2 and have rates of change that are changing at a rate of -2 ?
- a. $[A, B]$
 b. $[C, D]$
 c. $[E, F]$
 d. $[G, H]$
4. Which one of these intervals could g have a point of inflection?
- a. $[B, C]$
 b. $[D, E]$
 c. $[G, H]$
 d. The function g does not have a point of inflection between $[A, H]$



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web apr 22 2020 molar volume of gases 1 counting in moles one mole is defined as 6.02×10^{23} particles this is the avogadro number an unchanging constant whether you are counting hydrogen atoms carbon dioxide molecules or carbonate ions a mole is always 6.02×10^{23} but as different particles have different mass their molar mass varies

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web zero degrees celsius or 273 k is defined as standard temperature and one atmosphere or 760 mm hg is defined as standard pressure the volume occupied by one mole of a gas is called molar volume therefore the molar volume of a gas at standard temperature and pressure stp is 22.4 l

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