

CHEM 12: Hess's Law Worksheet

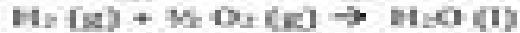
1. Calculate the ΔH for the reaction: $\text{C}_2\text{H}_6(\text{g}) + \text{H}_2(\text{g}) \rightarrow \text{C}_2\text{H}_4(\text{g})$, from the following data:



$$\Delta H = -1411 \text{ kJ / mole}$$



$$\Delta H = -1560 \text{ kJ / mole}$$



$$\Delta H = -285.8 \text{ kJ / mole}$$

2. Calculate the ΔH for the reaction:



from the following data:



$$\Delta H = -180.5 \text{ kJ / mole}$$



$$\Delta H = -91.8 \text{ kJ / mole}$$



$$\Delta H = -483.6 \text{ kJ / mole}$$

3. Find the ΔH_f° (heat of formation) for acetic acid, $\text{HC}_2\text{H}_3\text{O}_2$, using the following thermochemical data:



$$\Delta H = -875 \text{ kJ / mole}$$



$$\Delta H = -394.51 \text{ kJ / mole}$$



$$\Delta H = -285.8 \text{ kJ / mole}$$

Pure Factor C(s) is in the form of graphite.

Acetic acid, $\text{HC}_2\text{H}_3\text{O}_2$, actually has the structure: CH_3COOH



4. Calculate the ΔH for the reaction: $\text{CH}_4(\text{g}) + \text{NH}_3(\text{g}) \rightarrow \text{HCN}(\text{g}) + 3\text{H}_2(\text{g})$, from the following data:



$$\Delta H = -91.8 \text{ kJ / mole}$$



$$\Delta H = -74.9 \text{ kJ / mole}$$



$$\Delta H = +270.3 \text{ kJ / mole}$$

12 3 Hess S Law Answers

Kathleen Armour

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US Solutions Manual to Accompany Elements of Physical Chemistry 7e David Smith, 2017-09-28 The Solutions Manual to Accompany Elements of Physical Chemistry 7th edition contains full worked solutions to all end of chapter discussion questions and exercises featured in the book. The manual provides helpful comments and friendly advice to aid understanding. It is also a valuable resource for any lecturer who wishes to use the extensive selection of exercises featured in the text to support either formative or summative assessment and wants labour saving ready access to the full solutions to these questions

New Pattern NTA JEE Main Quick Guide in Chemistry with Numeric Answer Questions 3rd Edition

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Elements of Physical Chemistry Peter William Atkins, Julio De Paula, 2017 This revision of the introductory textbook of physical chemistry has been designed to broaden its appeal particularly to students with an interest in biological applications

Problems in Chemistry, Second Edition Daley, 1988-02-19 *Prentice Hall Chemistry*, 2000 *Physical Chemistry* Peter Atkins, Julio de Paula, Ronald Friedman, 2014

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ERISA Law Answer Book John F. Buckley, 2003 Taking a practical how to approach ERISA Law Answer Book places an emphasis on decision making as it focuses on questions commonly asked of pension and health plan advisors in connection with the administration of retirement programs. Rather than merely citing the Code and Regulations for authority and identifying cases which support the proposition or answer given this work sets forth a discussion of case authorities which have dealt with the relevant situation at hand and provides a careful analysis of how those cases apply. Key topics covered include Benefit arrangements subject to ERISA Construing and interpreting ERISA plans. Notice reporting and disclosure. Plan funding. Plan distribution. Benefit accrual and vesting. Fiduciary duties and

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