



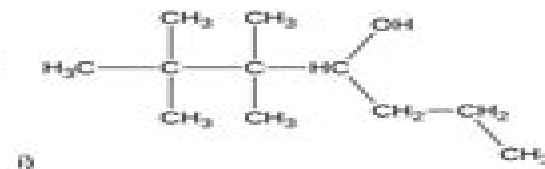
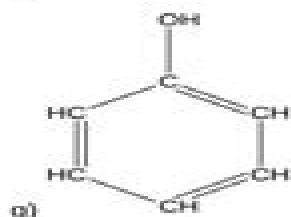
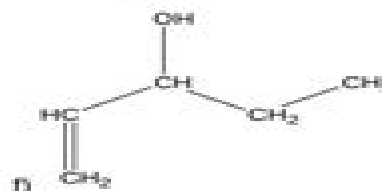
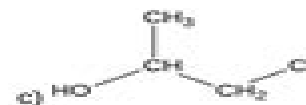
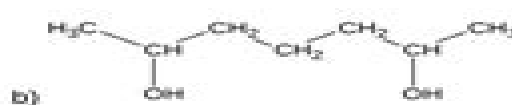
1. Draw the following alcohols

- a) heptan-2-ol
b) 3-methylhexan-1-ol
c) cyclopropanol

- d) 2,4,6-trichlorooctan-2-ol
e) pentan-1,4-diol
f) cyclohexene-1,3-diol

- g) but-2-ene-1-ol
h) 4-methylpent-2-yne-1-ol
i) 3,4-dimethylcycloheptan-1-ol

2. Name the following alcohols



3. Explain why the propane that is used as a fuel in a BBQ is gas at room temperature, but 2-propanol used as rubbing alcohol is a liquid at room temperature.

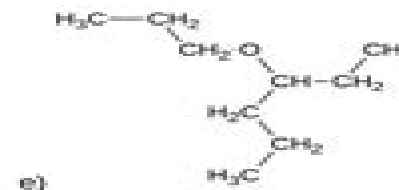
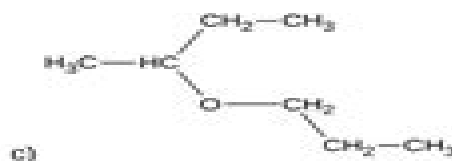
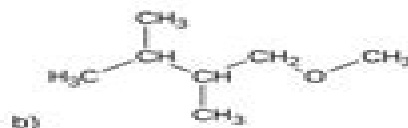
4. Draw the following ethers

- a) 1-propoxypentane
b) 2-ethoxybutane

- c) 1-methoxy-4-chlorohexane
d) 3-butoxy-2,4-dimethyloctane

- e) 2-propoxyheptane
f) 1-ethoxycyclopentane

5. Name the following ethers



22 Hydrocarbon Compounds Practice Problems Answer Key

Léo Pomar



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