

Stoichiometry Worksheet 2

1. Sodium chloride can be produced by the reaction of sodium and chlorine gas. Write a balanced equation for this reaction.

Suppose that 6.70 mol Na react with 3.20 mol Cl₂.

- What is the limiting reagent?
- How many moles of NaCl are produced?

2. The equation for the complete combustion of ethane, C₂H₆, is

$$\text{C}_2\text{H}_6 \text{ (g)} + \text{O}_2 \text{ (g)} \rightarrow \text{CO}_2 \text{ (g)} + \text{H}_2\text{O (g)}$$

Balance the equation. If 2.70 mol C₂H₆ is reacted with 6.30 mol O₂,

- identify the limiting reagent.
- calculate the moles of water produced.

3. Copper and sulfur combine to form copper (I) sulfide.

$$2 \text{ Cu (s)} + \text{S(s)} \rightarrow \text{Cu}_2\text{S (s)}$$

- What is the limiting reagent when 80.0 g Cu reacts with 25.0 g S?
- What is the maximum number of grams of Cu₂S that can be formed?

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A Stoichiometry Unit David Callaghan,2004 **Jacaranda Chemistry 2 VCE Units 3 and 4, 3e learnON and Print**
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Civil Engineering Review Manual Michael R. Lindeburg,1980 **The College Blue Book** ,1959 *The Software Finder* ,1983 *Annual Catalogue of the University of Kansas* Kansas. University,University of Kansas,1953 Introduction to Chemistry, Laboratory Manual T. R. Dickson,1994-12-23 Teaches chemistry by offering a dynamic provocative and relevant view of the topic and its importance to society and our daily lives Three themes are stressed throughout the text developing chemical thinking and a chemical vision learning problem solving methods and utilizing group work and discussion activities These themes involve and engage the students in their own learning processes they are challenged to be active The presentation of topics has been altered to include a new chapter which introduces the students to scientific thinking and shows that chemistry involves interesting and relevant topics The reorganization presents many core concepts in the first five chapters preparing students for later chapters In addition the author has added vignettes throughout the chapters referring to health technology the environment and society as well as to specific tools of direct use to students

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Nitrogen Control Risk Reduction Engineering Laboratory (U.S.),1994-12 This manual is constructed to progress from a broad discussion of nitrogen in the environment to the concepts using biological processes to control or remove nitrogen and finally to the details of designing specific systems

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