

Stoichiometry Worksheet 2

1. Sodium chloride can be produced by the reaction of sodium and chlorine gas. Write a balanced equation for this reaction.

Suppose that 6.70 mol Na react with 3.20 mol Cl_2 .

- What is the limiting reagent?
- How many moles of NaCl are produced?

2. The equation for the complete combustion of ethane, C_2H_6 , is



Balance the equation. If 2.70 mol C_2H_6 is reacted with 6.30 mol O_2 ,

- identify the limiting reagent.
- calculate the moles of water produced.

3. Copper and sulfur combine to form copper (I) sulfide.



- What is the limiting reagent when 80.0 g Cu reacts with 25.0 g S?
- What is the maximum number of grams of Cu_2S that can be formed?

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to Chemistry, Laboratory Manual T. R. Dickson, 1994-12-23 Teaches chemistry by offering a dynamic provocative and
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developing chemical thinking and a chemical vision learning problem solving methods and utilizing group work and
discussion activities These themes involve and engage the students in their own learning processes they are challenged to be
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in the first five chapters preparing students for later chapters In addition the author has added vignettes throughout the
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Prentice Hall Chemistry, 2000 An Investigation of the Manufacture of Water Gas with Especial Reference to the
Decomposition of Steam ... Lloyd Logan, 1929 **General Information and Announcements** University of Oklahoma, 1946
Nitrogen Control Risk Reduction Engineering Laboratory (U.S.), 1994-12 This manual is constructed to progress from a
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finally to the details of designing specific systems

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