

CHEVROLET TAHOE

2000-2006



SERVICE AND REPAIR MANUAL

2003 Chevy Tahoe Service Manual

JR Anderson

2003 Chevy Tahoe Service Manual:

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2 $\text{CO}_2 \text{ g} \rightarrow \text{H}_2 \text{O g}$ $\Delta h = 57 \text{ kJ}$
2 $\text{H}_2 \text{ g} \rightarrow \text{H}_2 \text{O g}$ $\Delta h = 94.5 \text{ kJ}$
 $\text{C}_2\text{H}_6 \text{ g} \rightarrow \text{H}_2 \text{O}_2 \text{ g}$ $\Delta h = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6 \text{ g} \rightarrow \text{CO}_2 \text{ g} + \text{H}_2 \text{ g}$ $\Delta h = 283 \text{ kJ}$
 $\text{C}_2\text{H}_6 \text{ g} \rightarrow \text{C}_2\text{H}_4 \text{ g} + \text{H}_2 \text{ g}$ $\Delta h = 1560 \text{ kJ}$

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 $\text{C}_2\text{H}_4 \text{ g} \rightarrow \text{CO}_2 \text{ g} + \text{H}_2 \text{O l}$ 1411 kJ mole
 $\text{C}_2\text{H}_6 \text{ g} \rightarrow \text{CO}_2 \text{ g} + \text{H}_2 \text{O l}$ 1560 kJ mole
 $\text{H}_2 \text{ g} \rightarrow \text{H}_2 \text{O l}$ 285 kJ mole

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web thus for reaction 9 2 in which 1 mole of H_3O and 1 mole of OH combine to form 2 moles of H_2O $\Delta H = 55.8 \text{ kJ}$ as shown note that the enthalpy change ΔH products reactants is positive if heat is absorbed that is if ΔH products

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