

Periodic Trends Lab

Periodic law states that when the elements are arranged in order of increasing atomic number, there is a periodic repetition of their chemical and physical properties. Elements within the same group have the most similar chemical and physical properties, although not identical. Elements in neighboring groups may also have similar chemical and physical properties, but to a lesser degree. Trends in properties occur within groups and across periods.

PURPOSE: In this activity, you will

- Interpret group and periodic trends in atomic radius, electronegativity, and ionization energy.
- Graph these trends.

PROCEDURE:

1. Complete the pre-lab section.
2. Using the data from **Table S** of your reference tables, prepare the graphs listed below. Staple your graphs to the back of this lab packet.

On each graph be sure to:

- **USE PENCIL!**
- Put an appropriate title and labels for each axis.
- Omit the transition elements.
- Connect the points you plot.
- Connect the points representing Group 1 elements with a green line.
- Connect the points representing Group 2 elements with a blue line.
- Connect the points representing Group 17 elements with a red line.
- Draw a triangle around the points representing the elements of Period 3.

GRAPH 1: Atomic radius vs. atomic number for elements 1-54.

GRAPH 2: Electronegativity vs. atomic number for elements 1-54.

GRAPH 3: Ionization energy vs. atomic number for elements 1-54.

3. Complete the conclusion section.

Answers To Periodic Trends Lab

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Hands-On General Science Activities With Real-Life Applications Pam Walker, Elaine Wood, 2008-04-21 In this second edition of Hands On General Science Activities with Real Life Applications Pam Walker and Elaine Wood have completely revised and updated their must have resource for science teachers of grades 5 12 The book offers a dynamic collection of classroom ready lessons projects and lab activities that encourage students to integrate basic science concepts and skills into everyday life

Laboratory Manual for Principles of General Chemistry Jo Allan Beran, 2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures

Exploring Physical Science in the Laboratory John T. Salinas , 2019-02-01 This full color manual is designed to satisfy the content needs of either a one or two semester introduction to physical science course populated by nonmajors It provides students with the opportunity to explore and make sense of the world around them to develop their skills and knowledge and to learn to think like scientists The material is written in an accessible way providing clearly written procedures a wide variety of exercises from which instructors can choose and real world examples that keep the content engaging Exploring Physical Science in the Laboratory guides students through the mysteries of the observable world and helps them develop a clear understanding of challenging concepts

Science Action Labs Physical Science Edward Shevick, 1998-03-01 Matter and Motion These easy to use hands on explorations are just what you need to get your science curriculum and your students into action

Foundations of Anatomy and Physiology - ePub Ellie Kirov, Alan Needham, 2023-04-01 This new practice manual is designed to provide students with the conceptual foundations of anatomy and physiology as well as the basic critical thinking skills they will need to apply theory to practice in real life settings

Written by lecturers Dr Ellie Kirov and Dr Alan Needham who have more than 60 years teaching experience between them the book caters to nursing health science and allied health students at varying levels of understanding and ability Learning activities are scaffolded to enable students to progress to more complex concepts once they have mastered the basics A key advantage of this manual is that it can be used by instructors and students in conjunction with any anatomy and or physiology core textbook or as a standalone resource It can be adapted for learning in all environments including where wet labs are not available Can be used with any other textbook or on its own flexible for teachers and students alike Scaffolded content suitable for students varying learning requirements and available facilities Concept based practical activities can be selected and adapted to align with different units across courses Provides a range of activities to support understanding and build knowledge including theory application and experimentation Activities can be aligned to learning requirements and needs may be selected to assist pre class in class post class or for self paced learning Easy to navigate icons identify content

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Integrated Science Laboratory Manual Michael J. Padilla,2000 Includes 74 investigations pre lab discussions and critical thinking questions safety manual and student safety test teaching support **Prentice Hall Chemistry** ,2000

Merrill Chemistry Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998 **The Essential Lab Manual** Karen Timberlake,2002-06-24 Drawing from the successful main Laboratory Manual the Essential Laboratory Manual includes twenty one experiments which have been revised and updated Suitable for a one or two term lab course Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction Experimental Chemistry Robert J. Artz,1982 The Software Encyclopedia 2001 ,2001 **Chemical Interactions** ,2005 **Chemistry** Eugene LeMay, Jr.,Herbert Beall,Karen M. Robblee,Douglas C. Brower,Prentice-Hall Staff,2002-02 **Chemistry** John S. Phillips,Cheryl Wistrom,2000 Programmed Learning and Individually Paced Instruction Carl H. Hendershot,1973 *EPIE Materials Report* ,1977 *World of Chemistry* Melvin D. Joesten,James L. Wood,1999 The quality of human life is described in terms understanding to the non science major in this application oriented chemistry textbook This paperback offers a more concise less theoretical less mathematical version of the hardback version WORLD OF CHEMISTRY Second Edition **Sage** ,1966

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