

## **Making Solutions Practice Worksheet**

- 1) Explain how you would make 450 mL of a 0.250 *M* NaOH solution.
  
  
  
  
  
  
  
  
  
  
- 2) To what volume will you have to dilute 30.0 mL of a 12 *M* HCl solution to make a 0.35 *M* HCl solution?
  
  
  
  
  
  
  
  
  
  
- 3) How many grams of calcium chloride will be needed to make 750 mL of a 0.100 *M* CaCl<sub>2</sub> solution?
  
  
  
  
  
  
  
  
  
  
- 4) Explain why this experimental procedure is incorrect: To make 1.00 L of a 1.00 *M* NaCl solution, I will dissolve 58.5 grams of sodium chloride in 1.00 L of water.

# Answers To Making Solutions Practice

**LP Steffe**



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*The Electrical Review*, 1874

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