

ASSIGNING OXIDATION NUMBERS

Name _____

Assign oxidation numbers to all of the elements in each of the compounds or ions below.

1. HCl	11. H_2SO_3
2. KNO_3	12. H_2SO_4
3. OH^-	13. BaO_2
4. Mg_3N_2	14. KMnO_4
5. KClO_3	15. LiH
6. $\text{Al}(\text{NO}_3)_3$	16. MnO_2
7. S_8	17. OF_2
8. H_2O_2	18. SO_3
9. PbO_2	19. NH_3
10. NaHSO_4	20. Na

Assigning Oxidation Numbers Answer Key

Ralph Thomas Myers

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Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter. Key revisions in this edition focus on three areas. The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry. Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering, biological, pharmaceutical and medical sciences. Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text. When students know what they know they are better able to learn and incorporate the material. Providing a total solution through New WileyPLUS by fully integrating the enhanced etext with online assessment, answer specific responses and additional practice resources. The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge. Problems are arranged in an intuitive confidence building order.

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