

## CHEM 101 ACS Study Guide

## Chapter 1

**SQ-1.** Planck's constant is  $6.626 \times 10^{-34}$  J-s. What is this value in kJ- $\mu$ s?

- (A)  $6.626 \times 10^{-43}$  kJ- $\mu$ s (B)  $6.626 \times 10^{-27}$  kJ- $\mu$ s  
(C)  $6.626 \times 10^{-34}$  kJ- $\mu$ s (D)  $6.626 \times 10^{-21}$  kJ- $\mu$ s

**SQ-8.** Twenty irregular pieces of an unknown metal have a collective mass of 28.225 g. When carefully placed in a graduated cylinder that originally contained 7.75 mL of water, the final volume read 11.70 mL. What is the density of the metal (in g-cm<sup>3</sup>)?

- (A) 2.41 g-cm<sup>-3</sup> (B) 3.64 g-cm<sup>-3</sup> (C) 7.15 g-cm<sup>-3</sup> (D) 11.2 g-cm<sup>-3</sup>

**PQ-12.** If a palladium nanoparticle has a density of 12.0 g-cm<sup>-3</sup>, what is the mass of a nanoparticle with a volume of  $1.84 \times 10^{-21}$  cm<sup>3</sup>?

- (A)  $2.21 \times 10^{-26}$  g (B)  $1.63 \times 10^{-26}$  g (C)  $6.52 \times 10^{-21}$  g (D)  $1.53 \times 10^{-22}$  g

**PQ-13.** Which sample has the largest volume?

Densities			
aluminum	2.7 g-cm <sup>-3</sup>	copper	9.0 g-cm <sup>-3</sup>

- (A) 1.0 g aluminum (B) 5.0 g aluminum (C) 1.0 g copper (D) 5.0 g copper

## Chapter 2

**SQ-3.** Which compound is named correctly?

- (A) FeCO<sub>3</sub>, iron carbonate (B) K<sub>2</sub>SO<sub>3</sub>, potassium sulfite  
(C) Sn(NO<sub>2</sub>)<sub>2</sub>, stannous nitrite (D) Co<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub>, cobalt(II) sulfate

# Acs Study Guide Answer Key

**June Eilers, Regina Fink**



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## **Table of Contents Acs Study Guide Answer Key**

1. Understanding the eBook Acs Study Guide Answer Key
  - The Rise of Digital Reading Acs Study Guide Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Acs Study Guide Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Acs Study Guide Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Acs Study Guide Answer Key
  - Personalized Recommendations
  - Acs Study Guide Answer Key User Reviews and Ratings
  - Acs Study Guide Answer Key and Bestseller Lists
5. Accessing Acs Study Guide Answer Key Free and Paid eBooks
  - Acs Study Guide Answer Key Public Domain eBooks
  - Acs Study Guide Answer Key eBook Subscription Services
  - Acs Study Guide Answer Key Budget-Friendly Options

6. Navigating Acs Study Guide Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Acs Study Guide Answer Key Compatibility with Devices
  - Acs Study Guide Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Acs Study Guide Answer Key
  - Highlighting and Note-Taking Acs Study Guide Answer Key
  - Interactive Elements Acs Study Guide Answer Key
8. Staying Engaged with Acs Study Guide Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Acs Study Guide Answer Key
9. Balancing eBooks and Physical Books Acs Study Guide Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Acs Study Guide Answer Key
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Acs Study Guide Answer Key
  - Setting Reading Goals Acs Study Guide Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Acs Study Guide Answer Key
  - Fact-Checking eBook Content of Acs Study Guide Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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