

Acids and bases review sheet Answer key

1. List the differences between an acid and a base. Your description should include such things as whether both are electrolytes, which contains a hydrogen ion and which includes a hydroxide ion, and you should also include two examples of an acid and two examples of a base that are present in your household.

Acid	base
Conducts electricity	conducts electricity
Contains hydrogen ion	contains hydroxide ion
Proton donor	proton acceptor
Turns litmus paper red	turns litmus paper blue
React with most metals	releases hydroxide ion
Releases hydronium	Ex: Windex
Ex: lemon	bleach
Vinegar	soap
Battery acid (sulfuric acid)	

2. Listed below are some of the properties of acids and bases. Fill in the blanks with the appropriate word, acids or bases:

- produce hydrogen ions (H^+) in solution
- have a bitter taste
- produce hydroxide ions (OH^-) ions
- react with bases to form salts
- have a sour taste
- have a slippery, soapy feel
- react with acids to form salts
- react with many metals

3. Write an equation for the dissociation of the following acids in water: follow example

- $HClO_4 \rightarrow H^+ + ClO_4^-$
- $H_2SO_4 \rightarrow 2H^+ + SO_4^{2-}$
- $HC_2H_5O_2 \rightarrow H^+ + C_2H_5O_2^-$
- $H_2S \rightarrow 2H^+ + S^{2-}$
- $HCl \rightarrow H^+ + Cl^-$
- $HNO_3 \rightarrow H^+ + NO_3^-$

4. Write an equation for the dissociation (ionization) of the following bases in water: (follow example)

- $Mg(OH)_2 \rightarrow Mg^{2+} + 2OH^-$
- $NaOH \rightarrow Na^+ + OH^-$
- $Ba(OH)_2 \rightarrow Ba^{2+} + 2(OH^-)$
- $KOH \rightarrow K^+ + OH^-$
- $Mg(OH)_2 \rightarrow Mg^{2+} + 2(OH^-)$
- $LiOH \rightarrow Li^+ + OH^-$

5) A. What type of reaction is a neutralization reaction? Double replacement

B. Write the net equation for neutralization:



Acid Base Review Worksheet Answer Key

Nurse Sarah

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1. Understanding the eBook Acid Base Review Worksheet Answer Key
 - The Rise of Digital Reading Acid Base Review Worksheet Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Acid Base Review Worksheet Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Acid Base Review Worksheet Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Acid Base Review Worksheet Answer Key
 - Personalized Recommendations
 - Acid Base Review Worksheet Answer Key User Reviews and Ratings
 - Acid Base Review Worksheet Answer Key and Bestseller Lists

5. Accessing Acid Base Review Worksheet Answer Key Free and Paid eBooks
 - Acid Base Review Worksheet Answer Key Public Domain eBooks
 - Acid Base Review Worksheet Answer Key eBook Subscription Services
 - Acid Base Review Worksheet Answer Key Budget-Friendly Options
6. Navigating Acid Base Review Worksheet Answer Key eBook Formats
 - ePUB, PDF, MOBI, and More
 - Acid Base Review Worksheet Answer Key Compatibility with Devices
 - Acid Base Review Worksheet Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Acid Base Review Worksheet Answer Key
 - Highlighting and Note-Taking Acid Base Review Worksheet Answer Key
 - Interactive Elements Acid Base Review Worksheet Answer Key
8. Staying Engaged with Acid Base Review Worksheet Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Acid Base Review Worksheet Answer Key
9. Balancing eBooks and Physical Books Acid Base Review Worksheet Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Acid Base Review Worksheet Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Acid Base Review Worksheet Answer Key
 - Setting Reading Goals Acid Base Review Worksheet Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Acid Base Review Worksheet Answer Key
 - Fact-Checking eBook Content of Acid Base Review Worksheet Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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