

# EF Products, Inc.

## R134a Oil Spec Chart

CAR MAKE	COMPRESSOR MANUFACTURER	COMPRESSOR TYPE	ISO VISCOSITY	EF LIQUID PART #	EF AEROSOL PART #
	Sanden	SD7	100	480 / 492	310
Land Rover	Sanden	SD7	Ester 100	481 / 491	312
Lotus	Sanden	SD7	100	480 / 492	310
Mack	Sanden	SD	100	480 / 492	310
Mazda	Nippondenso		46	484 / 494	310L
	Nippondenso		150	483 / 493	310H
	Panasonic		46	484 / 494	310L
	Sanden		100	480 / 492	310
	Zexel		46	484 / 494	310L
Mercedes-Benz	Harrison	A6, R4	150	483 / 493	310H
	Nippondenso	6SE, 7SB, 10P, 10PA	46	484 / 494	310L
	Sanden	SD6V	46	484 / 494	310L
	York		Ester 100	481 / 491	312
Mitsubishi	Mitsubishi	FX80, FXA80, FX105			
		FX105VS, MSC105	46	484 / 494	310L
	Nippondenso	10P	46	484 / 494	310L
New Holland	Sanden	SD	100	480 / 492	310
Nissan	Atsugi	NVR140S	100	480 / 492	310
	Calsonic	V6	46	484 / 494	310L
Opel	Harrison	V5	150	483 / 493	310H
	Nippondenso	6CA	46	484 / 494	310L
	Sanden	7SB	46	484 / 494	310L
	Zexel	KC45, KC50	100	480 / 492	310
Peterbilt	Sanden	SD	100	480 / 492	310
Peugeot	Sanden	SD	100	480 / 492	310
	Sanden	SDV	46	484 / 494	310L
Porsche	Nippondenso	6P, 6E, 7SB, 10P, 10PA	46	484 / 494	310L

# Ac Compressor Oil Capacity Chart

**DJ Losen**



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web solution average atomic mass of li 6 941 amu x 6 0151 amu 1 x 7 0160 amu 6 941 1 0009x 7 0160 1 0009x 0 075 x 0 075 x 0 075 corresponds to a natural abundance of 6li of 7 5 percent the natural abundance of 7li is x 0 925 or 92 5 percent 3 7 the unit factor required is 6 022

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web a molar mass li2co3 2 6 941 g 12 01 g 3 16 00 g 73 89 g b molar mass cs2 12 01 g 2 32 07 g 76 15 g c molar mass chcl3 12 01 g 1 008 g 3 35 45 g 119 37 g d molar mass c6h8o6 6 12 01 g 8 1 008 g 6 16 00 g 176 12 g e molar mass kno3 39 10 g 14 01 g 3 16 00 g 101 11 g

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web it is a small step from mole mass calculations to mass mass calculations if we start with a known mass of one substance in a chemical reaction instead of a known number of moles we can calculate the corresponding masses of other substances in the reaction

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web jul 4 2022 to calculate the mass of titanium metal that you can obtain multiply the number of moles of titanium by the molar mass of titanium  $47.867 \text{ g/mol}$  moles ti mass ti times molar mass ti  $4.12 \text{ mol ti} \times 47.867 \text{ g/mol ti} = 197 \text{ g ti}$  notag

chapter 3 mass relationships in chemical reactions - Nov 23 2022

web 41 42 chemistry ch 3 mass relationships in chemical reactions molecular mass of  $\text{NO}_2$  atomic mass of n 2 atomic mass of o 14 01 amu  $2 \times 16.00 \text{ amu} = 32.00 \text{ amu}$  atomic mass of  $\text{N}_2\text{O}_5$  2 atomic mass of n 5 atomic mass of o 2 14 01 amu  $5 \times 16.00 \text{ amu} = 80.00 \text{ amu}$  example 3 2 exercise 3 3