

## Nuclear Decay

### Types of Radiation

1. Complete the table below to compare the properties of alpha, beta, and gamma radiation.

Name	Alpha	Beta	Gamma
Greek letter	$\alpha$	$\beta$ -	$\gamma$
Symbol	${}^4_2\text{He}$	${}^0_{-1}e$	$\gamma$
Composition	2 protons, 2 neutrons	an electron	radiation (light)
Charge	0	-1	0
Stopped by	paper, skin, clothing	aluminum	lead

### Nuclear Decay Reactions

In problems 2 – 5, fill in the missing particle for each nuclear decay reaction. Then, match each equation to the appropriate type of nuclear decay.

- \_\_\_\_\_ 2.  ${}^{14}_6\text{C} \rightarrow {}^{14}_7\text{N} + \text{_____}$  beta A. alpha emission
- \_\_\_\_\_ 4.  ${}^{218}_{84}\text{Po} \rightarrow {}^{214}_{82}\text{Pb} + \text{_____}$  alpha B. beta emission
- \_\_\_\_\_ 5.  $\text{_____} + {}^0_{-1}e \rightarrow {}^{147}_{60}\text{Nd}$  electron capture C. electron capture

In problems 6 and 7, write a balanced equation for each nuclear decay reaction.

6. Decay of radium-223 by alpha ( $\alpha$ ) emission.



7. Decay of iodine-126 by electron ( $\beta$ ) emission.



8. How are the atomic number and mass number of a nuclide affected by Beta Decay?

The mass number does not change but the atomic number increases by 1.

9. How are the atomic number and mass number of a nuclide affected by the emission of gamma radiation?

The atomic number and mass number do not change.

10. Which has a slightly greater mass in a nuclear reaction, the reactants or the products? Why?

It depends on the type of reaction. A fusion reaction will have greater mass in the products because it is combining two nuclei. A fission reaction will not change mass.

# 2radioactive Decay Worksheet Answer Key

**Yijin Wang**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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Gardner,1957 *Radioactive Decay* ,1965 This is one in a series of SMSG supplementary and enrichment pamphlets for high school students This series is designed to make material for the study of topics of special interest to students readily accessible in classroom quantity Topics covered include the law of decay relative rate of change and a general solution MP

**Radioactive Decay Correction Tables** Owen H. Wheeler,Anne S. Kay,J. Elisa Trabal,1967 **Decay Schemes of Radioactive Nuclei** A B. S. Dzelepov,L. K. Peker,1966 *Radioactive Decay Tables* D. C. Kocher,1981 **Decay Scheme of  $^{45}\text{Rh106}$**  W. C. Peacock,1947 A General Algorithm for Radioactive Decay with Branching and Loss from a Medium

,1995 Many areas in the field of health physics require evaluation of the change of radionuclide quantity in a medium with time A general solution to first order compartmental models is presented in this paper for application to systems consisting of one physical medium that contains any number of radionuclide decay chain members The general analytical solution to the problem is first described mathematically and then extended to four applications 1 evaluation of the quantity of radionuclides as a function of time 2 evaluation of the time integral of the quantity during a time period 3 evaluation of the amount in a medium as a function of time following deposition at a constant rate and 4 evaluation of the time integral of the amount in a medium after deposition at a constant rate for a time The solution can be applied to any system involving physical transfers from the medium and radioactive chain decay with branching in the medium The general solution is presented for quantities expressed in units of atoms and activity Unlike many earlier mathematical solutions this solution includes chain decay with branching explicitly in the equations     Fallout from Nuclear Weapons Tests United States. Congress. Joint Committee on Atomic Energy,1959     **The Application of Coincidence Techniques in the Investigation of Radioactive Decay** J. M. Hill,1919     **RADIOACTIVE DECAY DATA TABLES - A HANDBOOK OF DECAY DATA FOR APPLICATION TO RADIATION DOSIMETRY AND RADIOLOGICAL ASSESSMENTS.** United States. Department of Energy. Technical Information Center,1981     *A Simplified Procedure for Computing the Growth of Radioactive Decay Products* Harold W. Kirby,D. A. Kremer,1954

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