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## Chemistry 11

## Classifying Matter

## Assignment

1. Complete the following sentences by filling in the appropriate word from the list below.

Elements  
Mixtures  
Compound  
Symbols  
Properties

Carbon  
Pure  
Letters  
Carbon dioxide

- a. Elements cannot be separated by physical or chemical means.
- b. Elements are abbreviated with symbols, which consist of one or two letters.
- c. Two or more elements chemically combined make up a compound.
- d. Carbon is an example of an element and carbon dioxide is an example of a compound.
- e. Elements and compounds are called pure substances because they have a unique set of chemical and physical properties.

2. Fill in the blanks in the following chart.

| Element Name | Element Symbol | Common Ions Formed |
|--------------|----------------|--------------------|
| a. Sodium    | <u>Na</u>      | <u>+1</u>          |
| b. Copper    | <u>Cu</u>      | <u>+2, +1</u>      |
| c. Tungsten  | <u>W</u>       | <u>+6</u>          |
| d. Iron      | <u>Fe</u>      | <u>+3, +2</u>      |
| e. Tin       | <u>Sn</u>      | <u>+4, +2</u>      |
| f. Potassium | <u>K</u>       | <u>+1</u>          |
| g. Gold      | <u>Au</u>      | <u>+3, +1</u>      |
| h. Silver    | <u>Ag</u>      | <u>+1</u>          |

3. Classify each of the following as one of an Atom (A), a Molecule (M) or an Ion (I).

a.  $\text{P}^{3-}$  I

b.  $\text{H}_2\text{O}$  M

c.  $\text{O}_2$  M

d.  $\text{Au}^{3+}$  I

e. Ge A

f. O A

g.  $\text{Ca}^{2+}$  I

h.  $\text{NH}_3$  M

# Answers For Classifying Matter Work Sheet

**Orit Hazzan, Tami Lapidot, Noa Ragonis**

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