

## SOLUBILITY CURVE WORKSHEET KEY

Use your solubility curve graphs provided to answer the following questions.

- What are the customary units of solubility on solubility curves? Degress Celsius and grams of solute/100g of water
- Define solubility. A measure of how much solute can dissolve in a given amount of solvent.
- According to the graph, the solubility of any substance changes as temperature changes.
- List the substances whose solubility decreases as temperature increases.  $\text{NH}_3$  and  $\text{Ce}_2(\text{SO}_4)_3$
- Which substance is least affected by temperature changes?  $\text{NaCl}$
- How many grams of ammonium chloride ( $\text{NH}_4\text{Cl}$ ) at  $50^\circ\text{C}$ ? 50g
- $\text{NaCl}$  and  $\text{KClO}_3$  have the same solubility at approximately  $78^\circ\text{C}$ .
- Which compound is least soluble in water at  $10^\circ\text{C}$ ?  $\text{KClO}_3$
- How many grams of  $\text{KNO}_3$  can be dissolved at  $50^\circ\text{C}$ ? 80g
- Are the following solutions unsaturated, saturated, or supersaturated?
  - 45g of  $\text{NaNO}_3$  in 100 g of water at  $30^\circ\text{C}$ . saturated
  - 60g of  $\text{KClO}_3$  in 100 g of water at  $90^\circ\text{C}$ . supersaturated
- How many grams of sodium chloride,  $\text{NaCl}$  are required to saturate 100 grams of water at  $100^\circ\text{ C}$ ? 40g
- How many grams of  $\text{NaNO}_3$  are required to saturate 100 grams of water at  $75^\circ\text{C}$ ? 140g
- How many grams of  $\text{KI}$  will saturate water at  $20^\circ\text{C}$ ? 33g
- At what temperature would 25g of potassium chlorate ( $\text{KClO}_3$ ) dissolve? 60°C
- At what temperature would 60g of  $\text{NH}_4\text{Cl}$  dissolve? 70°C
- 89 g  $\text{NaNO}_3$  is prepared at  $30^\circ\text{C}$ .
  - Will all of the salt dissolve? No
  - What mass of  $\text{NaNO}_3$  will dissolve at this temperature? 95g
- If 50 grams of  $\text{NH}_4\text{Cl}$  is dissolved at  $50^\circ\text{C}$ , how many additional grams  $\text{NH}_4\text{Cl}$  would be needed to make the solution saturated at  $80^\circ\text{C}$ ? 15g
- At  $50^\circ\text{C}$ , how many grams of  $\text{KNO}_3$  will dissolve? 80g
- At  $70^\circ\text{C}$ , how many grams of cerium (III) sulfate ( $\text{Ce}_2(\text{SO}_4)_3$ ) dissolve? 5g
- Determine if each of the following is unsaturated, saturated, or supersaturated.

a. 55g of $\text{NH}_3$ at $20^\circ\text{C}$ <u>supersaturated</u>	f. 78g of $\text{NaNO}_3$ at $10^\circ\text{C}$ . <u>saturated</u>
b. 10g of $\text{Ce}_2(\text{SO}_4)_3$ at $10^\circ\text{C}$ <u>unsaturated</u>	g. 145g of $\text{NaNO}_3$ at $80^\circ\text{C}$ . <u>saturated</u>
c. 110g of $\text{KNO}_3$ at $60^\circ\text{C}$ . <u>supersaturated</u>	h. 35g of $\text{NaCl}$ at $100^\circ\text{C}$ . <u>unsaturated</u>
d. 65g of $\text{NH}_4\text{Cl}$ at $80^\circ\text{C}$ . <u>saturated</u>	
e. 12g of $\text{NH}_3$ at $90^\circ\text{C}$ . <u>supersaturated</u>	

# Answer Key To Solubility Curve 1

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