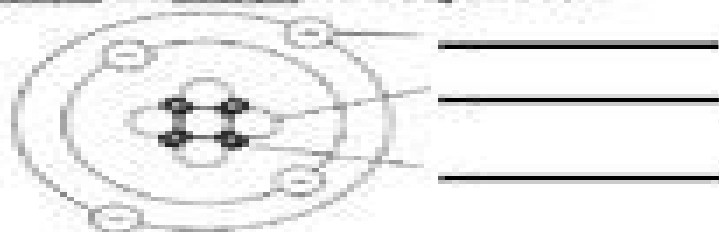


# Chemistry of Life Review

1. Complete the table below:

Name	Symbol	Atomic Number	Number of Protons	Number of Electrons	Number of Neutrons
Carbon					
	O				
Hydrogen					
Nitrogen					

2. Label the protons, neutrons, and electrons in the diagram below.



3. Match each word with the definition.

- |   |              |
|---|--------------|
| _____ 1. Anything that occupies space and has mass.                             | A. acid      |
| _____ 2. A substance made of two or more elements.                              | B. base      |
| _____ 3. Pairs off an element with different numbers of neutrons.               | C. carbon    |
| _____ 4. The smallest amount of a substance.                                    | D. catalyst  |
| _____ 5. A homogeneous mixture of one or more solutes dissolved in a solvent.   | E. compound  |
| _____ 6. A substance into which another substance dissolves to form a solution. | F. inorganic |
| _____ 7. A substance being dissolved in a mixture or solution.                  | G. isotopes  |
| _____ 8. A measure of acidity/basicity of a substance.                          | H. molecule  |
| _____ 9. A substance with weak attraction.                                      | I. organic   |
| _____ 10. A substance with four H+ ions.  | J. pH        |
| _____ 11. A carbon-based substance in living systems.                           | K. solute    |
| _____ 12. A substance that does not contain carbon.                             | L. solution  |
| _____ 13. The basic element in organic compounds.                               | M. solvent   |
| _____ 14. A substance that speeds up a chemical reaction.                       | N. cellulose |
| _____ 15. A polysaccharide in plants that serves as building material.          | O. matter    |

# Answer Key To The Chemistry Of Life

**Carolyn Tarrant, Eva Maria  
Krockow, Emily Kate Rousham, Björn  
Rönnerstrand**



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