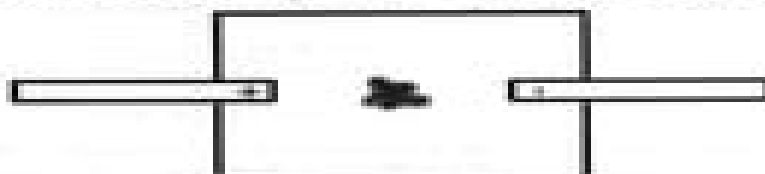


9. A mixture of  $\text{KCl}$  and  $\text{KCHO}_2$  weighing 1.8 gms was heated. After the reaction, the dry oxygen gas generated occupied 140 ml at STP. What percent of original mixture was  $\text{KCHO}_2$ ?
- 28.4%
  - 37.5%
  - 42.6%
  - 64.8%
10. The  $\text{pK}_a$  of Xanthine monophosphate (XMP) is 5.5. A certain experiment can detect all forms of a solute molecule present in solution at greater than 10% of the total solute concentration. At pH 7, the experiment will detect presence of:
- Neutral XMP and deprotonated XMP.
  - Deprotonated XMP
  - Neutral XMP
  - None of the above
11. Using molecular orbital theory or otherwise, arrange the molecules  $\text{KO}_2$ ,  $\text{O}_2$ , and  $\text{O}_2[\text{AsF}_6]$  in decreasing order of the O-O bond length
- $\text{KO}_2 > \text{O}_2 > \text{O}_2[\text{AsF}_6]$
  - $\text{O}_2 > \text{O}_2[\text{AsF}_6] > \text{KO}_2$
  - $\text{O}_2[\text{AsF}_6] > \text{O}_2 > \text{KO}_2$
  - $\text{KO}_2 > \text{O}_2[\text{AsF}_6] > \text{O}_2$
12. In an experiment 'E-A' potassium chromate was placed in the centre of a wet filter paper and electrodes were connected as shown. Which of the following do you expect to observe?



- Yellow color at positive terminal
- Purple color at negative terminal
- Yellow color at negative terminal
- Purple color at positive terminal

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**Peipei Pang**



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